

NEW FARNESYLACETONE DERIVATIVES FROM SARGASSUM MICRACANTHUM

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From the brown alga, Sargassum micracanthum, eight new farnesyl-  
acetone derivatives were isolated and their structures were elucidated  
from spectral and chemical properties.

Many species of the brown algae of Sargassum family are known to contain prenyl  
phenyl derivatives,<sup>1</sup> methylhydroquinones and methylbenzoquinones bearing geranyl-  
geranyl side chain. In the course of our investigations on the constituents of  
Sargassum family growing on the Japanese coast, we isolated new terpenoids which  
biogenetically seemed to be derived from these prenyl phenyl derivatives.

The methanol extracts of fresh S. micracanthum (Kützting) Yendo,<sup>2</sup> togemoku in  
Japanese, were chromatographed on silica gel, affording three fractions, I-III. The  
fraction I consisted of almost pure farnesylacetone (1); IR(CCl<sub>4</sub>) 1715 cm<sup>-1</sup>; NMR  
(CCl<sub>4</sub>) δ 1.62 (9H,s), 1.68 (3H,s), 2.08 (3H,s), 5.08 (3H,br.t), which was identified  
by converting into the semicarbazone; mp 79-80° (lit.<sup>3</sup> mp 80.5-81.5°). Although the  
TLC of the fraction II showed only one spot, its HPLC exhibited five distinct peaks,  
and four new compounds (2-5) were obtained from this fraction by preparative HPLC  
(TSK-GEL LS-410, 7.5 mm x 30 cm column, CH<sub>3</sub>CN : H<sub>2</sub>O = 3:1). The fraction III was a  
mixture of four compounds (7-10) which was separated by preparative TLC and HPLC.

The molecular formula, C<sub>18</sub>H<sub>30</sub>O<sub>2</sub> (M<sup>+</sup> 278.2230, calcd 278.2245), of COMPOUND 2,  
corresponded to that of dihydromonooxofarnesylacetone. The presence of -COCH=C(CH<sub>3</sub>)  
moiety and an acetyl group was deduced from the IR absorption bands at 1680 and 1715  
cm<sup>-1</sup>, the UV absorption maximum at 239 nm (ε 10500), and the NMR signals at δ 1.85  
(3H,d,J=1.5 Hz), 5.95 (1H,br.s), and 2.07 (3H,s). The NMR spectrum also suggested  
the presence of isopropyl [δ 0.92 (6H,d,J=7 Hz)], vinyl methyl [1.62 (3H,s)], and  
trisubstituted olefin [5.10 (1H,br.t,J=7 Hz)] groups. The stereochemistry of the

double bonds at C-5 and C-10 was determined to be E and Z, respectively, from the chemical shifts of the methyl groups at C-6 ( $\delta$  1.62) and C-10 ( $\delta$  1.85).<sup>4</sup> The structure was further verified by high resolution mass spectrum as described in the Figure. Hydrogenation of 2 afforded the tetrahydro derivative (6), C<sub>18</sub>H<sub>34</sub>O<sub>2</sub>; m/e 282(M<sup>+</sup>), 267(M<sup>+</sup>-CH<sub>3</sub>), 225(M<sup>+</sup>-C<sub>4</sub>H<sub>9</sub>, M<sup>+</sup>-CH<sub>2</sub>COCH<sub>3</sub>), 182 (M<sup>+</sup>-Me<sub>2</sub>CHCH<sub>2</sub>COCH<sub>3</sub>); IR(CCl<sub>4</sub>) 1715 cm<sup>-1</sup>.

COMPOUND 3, C<sub>18</sub>H<sub>32</sub>O<sub>2</sub>; m/e 280.2378(M<sup>+</sup>), 262 (M<sup>+</sup>-H<sub>2</sub>O), 223 (M<sup>+</sup>-C<sub>4</sub>H<sub>9</sub>, M<sup>+</sup>-CH<sub>2</sub>COCH<sub>3</sub>) was a dihydro derivative of compound 2. The IR spectrum (1710 cm<sup>-1</sup>) showed the absence of an  $\alpha,\beta$ -unsaturated carbonyl group, and the NMR spectrum revealed the presence of an acetyl ( $\delta$  2.05, 3H, s), isopropyl (0.90, 6H, d, J=7 Hz), and secondary methyl (0.87, 3H, d, J=7 Hz) groups, together with a trisubstituted olefin group (5.02, 1H, m). The structure 3 was confirmed by converting it into the diketone (6) by hydrogenation.

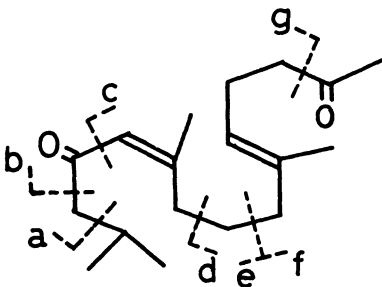
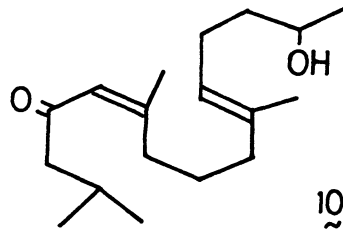
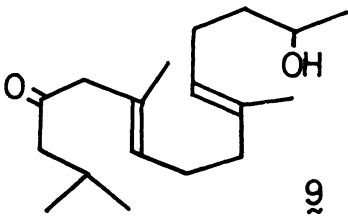
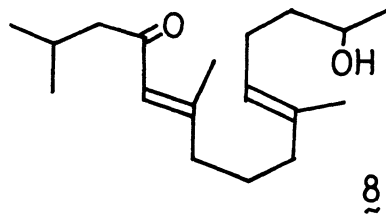
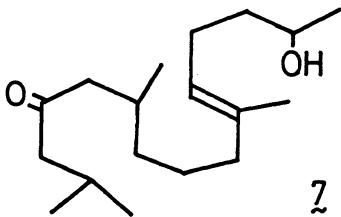
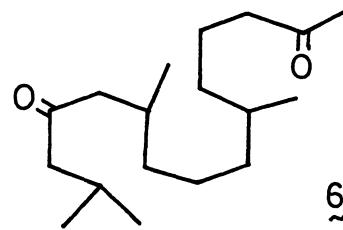
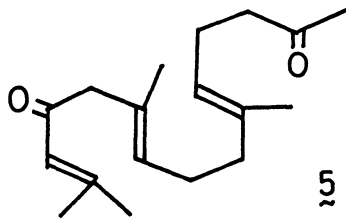
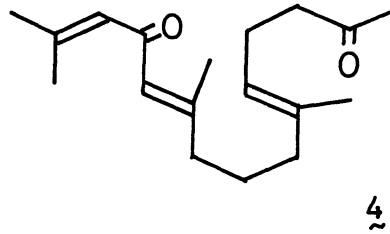
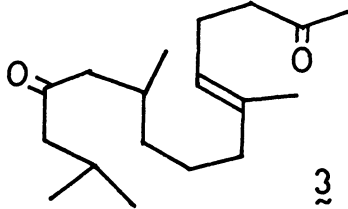
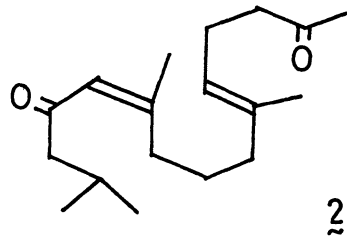
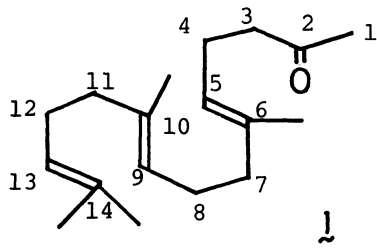
COMPOUND 4, C<sub>18</sub>H<sub>28</sub>O<sub>2</sub>; m/e 276.2141(M<sup>+</sup>) and 219 (M<sup>+</sup>-CH<sub>2</sub>COCH<sub>3</sub>), showed a strong absorption maximum at 246 nm ( $\epsilon$ 20900) in its UV spectrum, and exhibited IR absorption bands due to  $\alpha,\beta$ -unsaturated carbonyl system at 1670 and 1625 cm<sup>-1</sup>. The molecular formula corresponded to that of dehydro derivative of compound 2, and the presence of an isopropylidene group instead of isopropyl group was confirmed by its NMR spectrum;  $\delta$  1.62 (3H, s, 6-Me), 1.86 (3H, s, 14-Me, trans to C=O), 2.05 (3H, s, COCH<sub>3</sub>), 2.11 (6H, d, J=1 Hz, 10-Me and 14-Me, cis to C=O), 5.02 (1H, t, J=7 Hz, 5-H), and 5.90 (2H, m, 11-H, and 13-H). The E-configuration of the double bond at C-10 was obvious from the down-field chemical shift ( $\delta$  2.11) of the methyl group at C-10.

COMPOUND 5, C<sub>18</sub>H<sub>28</sub>O<sub>2</sub>, was an isomer of 4, and showed the following spectral properties; m/e 276.2042 (M<sup>+</sup>), 261 (M<sup>+</sup>-CH<sub>3</sub>), 233 (M<sup>+</sup>-COCH<sub>3</sub>), 83 (Me<sub>2</sub>C=CHCO); IR(CCl<sub>4</sub>) 1715, 1685, 1615 cm<sup>-1</sup>; NMR(CCl<sub>4</sub>)  $\delta$  1.62 (6H, s), 1.87 (3H, s), 2.04 (6H, s), 2.93 (2H, s), 5.0-5.3 (2H, m), 6.01 (1H, s). The stereochemistry of the double bonds at C-5 and C-9 was determined to be E from the chemical shifts of the methyl signals.

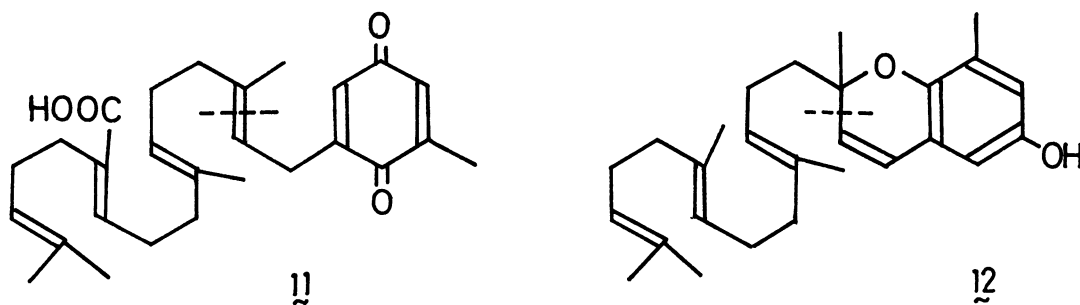
The structures 7-10 were assigned for the remaining new compounds on the basis of the following spectral properties.

COMPOUND 7, C<sub>18</sub>H<sub>34</sub>O<sub>2</sub>; m/e 282.2561 (M<sup>+</sup>), 264 (M<sup>+</sup>-H<sub>2</sub>O), 225 (M<sup>+</sup>-C<sub>4</sub>H<sub>9</sub>), 85 (Me<sub>2</sub>CH-CH<sub>2</sub>CO); IR(CCl<sub>4</sub>) 3620, 1710, 1615 cm<sup>-1</sup>; NMR(CCl<sub>4</sub>)  $\delta$  0.87 (3H, d, J=7 Hz), 0.92 (6H, d, J=7 Hz), 1.13 (3H, d, J=7 Hz), 3.70 (1H, m), 5.10 (1H, m), 1.62 (3H, s).

COMPOUND 8, C<sub>18</sub>H<sub>32</sub>O<sub>2</sub>; m/e 280.2393 (M<sup>+</sup>), 265 (M<sup>+</sup>-CH<sub>3</sub>), 223 (M<sup>+</sup>-C<sub>4</sub>H<sub>9</sub>), 85 (Me<sub>2</sub>CH-CH<sub>2</sub>CO);  $[\alpha]_D$  -3.75° (chloroform, c 0.53);  $\lambda_{\max}^{\text{EtOH}}$  241 nm ( $\epsilon$  9500); IR(CCl<sub>4</sub>) 3620, 1685, 1620, 1110, 1050 cm<sup>-1</sup>; NMR(CCl<sub>4</sub>)  $\delta$  0.92 (6H, d, J=7 Hz), 1.13 (3H, d, J=7 Hz), 1.61 (3H, s),



Fragment	Found	Calcd	Formula
a	235.1667	.1698	C <sub>15</sub> H <sub>23</sub> O <sub>2</sub>
b	221.1545	.1541	C <sub>14</sub> H <sub>21</sub> O <sub>2</sub>
c	193.1581	.1592	C <sub>13</sub> H <sub>21</sub> O
d	140.1198	.1201	C <sub>9</sub> H <sub>16</sub> O
e	153.1257	.1279	C <sub>10</sub> H <sub>17</sub> O
f	125.0950	.0966	C <sub>8</sub> H <sub>13</sub> O
g	235.2029	.2061	C <sub>16</sub> H <sub>27</sub> O



2.09 (3H,s), 3.70 (1H,sextet,J=7 Hz), 5.10 (1H,t,J=7 Hz), 5.94 (1H,s).

COMPOUND 9,  $C_{18}H_{32}O_2$ ;  $m/e$  280.2360 ( $M^+$ ), 262 ( $M^+-H_2O$ ), 247 (262- $CH_3$ ), 237 ( $M^+-C_3H_7$ ), 223 ( $M^+-C_4H_9$ ), 85 (base,  $Me_2CHCH_2CO$ );  $[\alpha]_D -3.31^\circ$  (chloroform,  $c$  0.45); IR( $CCl_4$ ) 3620, 1710  $cm^{-1}$ ; NMR( $CCl_4$ )  $\delta$  0.90 (6H,d,J=7 Hz), 1.12 (3H,d,J=7 Hz), 1.61 (6H,s), 2.93 (2H,s), 3.73 (1H,sextet,J=7 Hz), 5.16 (2H,m).

COMPOUND 10,  $C_{18}H_{32}O_2$ ;  $m/e$  280.2383 ( $M^+$ ), 262 ( $M^+-H_2O$ ), 223 ( $M^+-C_4H_9$ ), 85 ( $Me_2CH-CH_2CO$ );  $[\alpha]_D -3.16^\circ$  (chloroform,  $c$  0.25),  $\lambda_{max}^{EtOH}$  239 nm ( $\epsilon$  10600), IR( $CCl_4$ ) 3620, 1685, 1615, 1170, 1150  $cm^{-1}$ ; NMR( $CCl_4$ )  $\delta$  0.92 (6H,d,J=7 Hz), 1.13 (3H,d,J=7 Hz), 1.64 (3H,s), 1.85 (3H,s), 3.68 (1H,sextet,J=7 Hz), 5.10 (1H,t), 5.90 (1H,s).

From the brown algae belonging to Sargassum group, geranylgeranylquinones and chromenols such as sargaquinoic acid (11)<sup>1c</sup> and  $\delta$ -tocotrienol (12)<sup>1a</sup> have been isolated. The farnesylacetone derivatives reported here are supposed to be "norditerpene"s, derived from these quinones and chromenols by the oxidative cleavage of the C-C bonds pointed by the dotted lines in the Figure.

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